
	QMRF identifier (JRC Inventory):	
	QMRF Title: <i>QSAR model for Persistence: Abiotic degradation in air</i>	
	Printing Date: <i>14.06.2010</i>	

1. QSAR identifier

1.1. QSAR identifier (title):

QSAR model for Persistence: Abiotic degradation in air

1.2. Other related models:

1.3. Software coding the model:

QSARModel 4.0.4 Molcode Ltd., Turu 2, Tartu, 51014, Estonia
<http://www.molcode.com>

2. General information

2.1. Date of QMRF:

30.04.2010

2.2. QMRF author(s) and contact details:

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2.3. Date of QMRF update(s):

2.4. QMRF update(s):

2.5. Model developer(s) and contact details:

Molcode model development team Molcode Ltd. Turu 2, Tartu, 51014, Estonia models@molcode.com http://www.molcode.com

2.6.Date of model development and/or publication:

28.04.2010

2.7.Reference(s) to main scientific papers and/or software package:

[1]Karelson M, Dobchev D, Tamm T, Tulp I, Jänes J, Tämm K, Lomaka A, Savchenko D & Karelson G (2008). Correlation of blood-brain penetration and human serum albumin binding with theoretical descriptors. ARKIVOC 16, 38-60.

[2]Karelson M, Karelson G, Tamm T, Tulp I, Jänes J, Tämm K, Lomaka A, Savchenko D & Dobchev D (2009). QSAR study of pharmacological permeabilities. ARKIVOC 2, 218-238.

2.8.Availability of information about the model:

All information in full detail is available

2.9.Availability of another QMRF for exactly the same model:

None to date

3.Defining the endpoint - OECD Principle 1

3.1.Species:

n/a

3.2.Endpoint:

2.Environmental fate parameters 3.Persistence: Biodegradation

2.3.a.Ready/not ready biodegradability

3.3.Comment on endpoint:

The half-life is the time required for the concentration of a substance to halve its original value in a particular environmental medium. The half-lives of organic compounds are among the most commonly used criteria for studying persistence [1]. The semiquantitative data based on expert judgment and actual experimental values have already been suggested by Webster et al. [2] as preferable for half life identification, and are commonly used to develop the widely applied multimedia models [3,4]. In addition, a simple QSPR regression model has been demonstrated to be an useful tool for the identification and prioritization of existing or not yet synthesized potential persistent organic pollutants [5].

3.4.Endpoint units:

The half-life values (55-55000 h) were transformed into logarithmic form for modelling.

3.5.Dependent variable:

log T(0.5)

3.6.Experimental protocol:

The dataset of structurally heterogeneous and highly representative of many classes of already defined problematic chemicals includes 250 organic compounds of known half-lives for transformation into air [6].

3.7. Endpoint data quality and variability:

A collection of experimental data from different labs has been used; Semiquantitative degradation half lives in air were organized in nine half-life categories.

Statistics:

max value: 4.74

min value: 0.699

standard deviation: 0.832

skewness: 1.02

4. Defining the algorithm - OECD Principle 2

4.1. Type of model:

2D and 3D regression-based QSAR

4.2. Explicit algorithm:

multilinear regression QSAR

multilinear regression QSAR derived with BMLR (Best Multiple Linear Regression) method

$$\log T(0.5) = 0.273$$

-0.364*HOMO energy (AM1)

-0.219*Min valency (AM1) for O atoms

-4.085*Relative number of H atoms

+3.955E-002*count of H-donor sites (AM1)

-0.307*Square root of Charged (AM1) Surface Area of N atoms

4.3. Descriptors in the model:

[1]HOMO energy (AM1) eV Energy of highest occupied molecular orbital energy

[2]Min valency (AM1) for O atoms unitless minimum valency on oxygen atoms

[3]Relative number of H atoms unitless Relative number of H atoms

[4]count of H-donor sites (AM1) unitless count of H-donor sites (AM1)u

[5]Square root of Charged (AM1) Surface Area of N atoms Å Square root of Charged Surface Area of N atoms based on AM1 calculations

4.4. Descriptor selection:

Initial pool of ~1000 descriptors. Stepwise descriptor selection based on a set of statistical selection rules (one-parameter equations: Fisher criterion and R2 over threshold, variance and t-test value over threshold, intercorrelation with another descriptor not over threshold),

(two-parameter equations: intercorrelation coefficient below threshold, significant correlation with endpoint, in terms of correlation coefficient and t-test)

Stepwise trial of additional descriptors not significantly correlated to any already in the model.

4.5. Algorithm and descriptor generation:

1D, 2D, and 3D theoretical calculations. Quantum chemical descriptors derived from AM1 calculation. Model developed by using multilinear regression.

4.6. Software name and version for descriptor generation:

QSARModel 4.0.4

QSAR/QSPR package that will compute chemically meaningful descriptors and includes statistical tools for regression modeling

Molcode Ltd, Turu 2, Tartu, 51014, Estonia

<http://www.molcode.com>

4.7. Chemicals/Descriptors ratio:

33.4 (167 chemicals / 5 descriptors)

5. Defining the applicability domain - OECD Principle 3

5.1. Description of the applicability domain of the model:

Applicability domain based on training set:

a) by chemical identity: diverse set of organic pollutants (aromatic, aliphatic and cyclic amines, ketones, alcohols, esters, etc)

b) by descriptor value range: The model is suitable for compounds that have the descriptors

in the following minimal-maximal range:

HOMO energy (AM1): -12.2 - -7.92

Min valency (AM1) for O atoms: 0 - 2.39

Relative number of H atoms: 0.00 - 0.70

count of H-donor sites (AM1) : 0.00 - 19.0

Square root of Charged (AM1) Surface Area of N atoms: 0.00 - 3.60

5.2. Method used to assess the applicability domain:

Range of descriptor values in training set with $\pm 30\%$ confidence.

Descriptor values must fall between maximal and minimal descriptor values of training set $\pm 30\%$.

5.3. Software name and version for applicability domain assessment:

QSARModel 4.0.4

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5.4. Limits of applicability:

See 5.1

6. Internal validation - OECD Principle 4

6.1. Availability of the training set:

Yes

6.2. Available information for the training set:

CAS RN: Yes

Chemical Name: Yes

Smiles: No

Formula:Yes

INChI:No

MOL file:Yes

6.3.Data for each descriptor variable for the training set:

All

6.4.Data for the dependent variable for the training set:

All

6.5.Other information about the training set:

167 data points

0 negative values

167 positive values

6.6.Pre-processing of data before modelling:

n/a

6.7.Statistics for goodness-of-fit:

$R^2 = 0.714$ (Correlation coefficient)

$s^2 = 0.453$ (Standard error of the estimate)

$F = 80.3$ (Fisher function)

6.8.Robustness - Statistics obtained by leave-one-out cross-validation:

$R^2_{CV} = 0.692$

6.9.Robustness - Statistics obtained by leave-many-out cross-validation:

$R^2_{CVMO} = 0.689$

6.10.Robustness - Statistics obtained by Y-scrambling:

n/a

6.11.Robustness - Statistics obtained by bootstrap:

n/a

6.12.Robustness - Statistics obtained by other methods:

ABC analysis (2:1 training : prediction) on sorted (in increased order of endpoint value) data divided into 3 subsets (A;B;C). Training set formed with 2/3 of the compounds (set A+B, A+C, B+C) and validation set consisted of 1/3 of the compounds (C, B, A).

average R^2 (fitting) = 0.717

average R^2 (prediction) = 0.660

7.External validation - OECD Principle 4

7.1.Availability of the external validation set:

Yes

7.2.Available information for the external validation set:

CAS RN:Yes

Chemical Name:Yes

Smiles:No

Formula:Yes

INChI:No

MOL file:Yes

7.3.Data for each descriptor variable for the external validation set:

All

7.4.Data for the dependent variable for the external validation set:

All

7.5.Other information about the external validation set:

83 data points,

0 negative values,

83 positive values

7.6.Experimental design of test set:

From sorted source data, each 3rd was subjected to the test set.

7.7.Predictivity - Statistics obtained by external validation:

$R^2 = 0.602$ (Coefficient of determination)

7.8.Predictivity - Assessment of the external validation set:

Two compounds were leaved out becuae their descriptor "Square root of Charged (AM1) Surface Area of N atoms" values did not fitted into the applicability domain. These compounds are:2,4-dinitrotoluene (CAS: 121-14-2)4,6-dinitro-o-cresol (CAS: 534-52-1)

The rest are in range of applicability domain:

HOMO energy (AM1): -12.4 - -7.86

Min valency (AM1) for O atoms: 1.80 - 2.29

Relative number of H atoms: 0.0455 - 0.706

count of H-donor sites (AM1) : 0.00 - 21.0

Square root of Charged (AM1) Surface Area of N atoms: 0.00 - 3.70

7.9.Comments on the external validation of the model:

The validation coefficient of determination (R^2) is significant and close to those coefficients of internal validation (R^2_{CV} and R^2_{CVM}).

8.Providing a mechanistic interpretation - OECD Principle 5

8.1.Mechanistic basis of the model:

"HOMO energy (AM1)" reflects an ionization potential of molecule.

"Relative number of H atoms" indirectly shows a magnitude of unsaturated bonds (including aromaticity). Smaller number of hydrogens in carbon chain indicates larger content of unsaturated bonds. These bonds include p-orbitals which are important for carbon chain degradation. Hydrogen bonding capability is covered by descriptor "count of H-donor sites (AM1)". Others atom specific descriptors are counting reactivity of these heteroatom sites.

8.2.A priori or a posteriori mechanistic interpretation:

a posteriori mechanistic interpretation, consistent with published scientific interpretations of experiments [5]

8.3.Other information about the mechanistic interpretation:

9.Miscellaneous information

9.1.Comments:

The data are gathered from handbook (Physical-Chemical Properties and Environmental Fate Handbook) which includes data from different

sources. Therefore the experimental protocol cannot be provided. The data were also semiquantitatively classified as proposed by Mackay [1]. [1] Webster, E.; Mackay, D.; Wania, F. Evaluating Environmental Persistence. Environ. Toxicol. Chem. 1998, 17, 2148-2158.

9.2. Bibliography:

[1] UNEP, Stockholm Convention on Persistent Organic Pollutants, United Nations Environment Program, Geneva, Switzerland, 2001 <http://www.pops.int>

[2] Webster, E.; Mackay, D.; Wania, F. Evaluating Environmental Persistence, Environ. Toxicol. Chem. 1998, 17, 2148-2158.

[3] Klasmeier, J.; Matthies, M.; MacLeod, M.; Fenner, K.; Scheringer, M.; Stroebe, M.; Le Gall, A. C.; McKone, T.; Van De Meent, D.; Wania, F. Application of Multimedia Models for Screening Assessment of Long-Range Transport Potential and Overall Persistence, Environ. Sci. Technol. 2006, 40, 53-60.

[4] Fenner, K.; Scheringer, M.; Macleod, M.; Matthies, M.; McKone, T.; Stroebe, M.; Beyer, A.; Bonnell, M.; Le Gall, A. C.; Klasmeier, J.; Mackay, D.; Van de Meent, D.; Pennington, D.; Scharenberg, B.; Suzuki, N.; Wania, F. Comparing Estimates of Persistence And Long-Range Transport Potential among Multimedia Models, Environ. Sci. Technol. 2005, 39, 1932-1942.

[5] Gramatica, P.; Papa, E. Screening and ranking of POPs for global half-life: QSAR approaches for prioritization based on molecular structure, Environ. Sci. Technol. 2007, 41, 2833-2839.

[6] Mackay, D.; Shiu, W. Y.; Ma, K. C. Physical-Chemical Properties and Environmental Fate Handbook, CRCnet-BASE CD-ROM; Chapman and Hall/CRC: Boca Raton, FL, 2000.

9.3. Supporting information:

Training set(s) Test set(s) Supporting information

Karelson Arkivoc 2008	http://qsardb.jrc.it:80/qmrf/download_attachment.jsp?name=qmrf83_KarelsonArkivoc2008.pdf
Karelson Arkivoc 2009	http://qsardb.jrc.it:80/qmrf/download_attachment.jsp?name=qmrf83_KarelsonArkivoc2009.pdf

10. Summary (ECB Inventory)

10.1. QMRF number:

10.2. Publication date:

10.3. Keywords:

10.4. Comments: